# NEW GERMACRANOLIDES, GUAIANOLIDES AND REARRANGED GUAIANOLIDES FROM LASIOLAENA SANTOSII\*

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Abstract—The aerial parts of *L. santosii* afforded three new diterpenes, a phenylpropanol, three eudesmane derivatives, three germacranolides, seventeen guaianolides and four rearranged guaianolides of a, so far, unknown type. Furthermore, several known compounds were isolated. The structures of the new compounds, the separations of which caused difficulties, were elucidated by spectroscopic methods and by some chemical transformations. The chemotaxonomic situation is discussed briefly.

#### INTRODUCTION

The small Brazilian genus Lasiolaena [1] (Compositae, tribe Eupatorieae) is placed in the Gyptis group [2], closest to the widely distributed Conocliniopsis, which also has the receptacle highly conical but glabrous and the pappus setae narrowed at the tip. Agrianthus and Bahianthus are related genera. From the latter the widely distributed euparin derivatives and kolavanes have been isolated [3], while Agrianthus contains highly oxygenated guaianolides [4], and Conocliniopsis contains heliangolides with a furanone ring [5]. Conoclinium, also placed in this group, gave several flavones only [6]. None of the remaining genera of this group has been studied chemically. We have now investigated Lasiolaena santosii K. et R. to see whether the chemistry showed further indications of relationships in this diverse group. The aerial parts of this plant were found to be extremely rich in sesquiterpene lactones, among them four lactones with a new type of rearranged carbon skeleton. Furthermore, some new diterpenes and eudesmane derivatives, as well as an unusual phenylpropanol ester, were isolated.

### RESULTS AND DISCUSSION

The roots of *L. santosii* afforded squalene, dammadienone, dammadienol and its acetate, while the aerial parts, in addition to these compounds, contained germacrene D, caryophyllene epoxide, lupeol and its acetate, as well as the isomers 1–4 [7], the triacetate 5 [8] and a very complex mixture of polar compounds from which the germacranolides eupatolide (13) [9] and balchanolide (14) [10], the flavone 45 and thirty new compounds were isolated. The latter were made up of the *ent*-labdane derivative 6, the geranyl geraniol derivatives 7

and 8, the 7-epimeric methyl esters of  $1\beta$ -hydroxycostic acid 10 and 11, the eudesmanolide 12, the germacranolides 15-17, the rearranged guaianolides 21-24, the guaianolides 25-27 and 29-41, and the phenyl propanol derivative 43. The latter is closely related to the corresponding dimethoxy compound isolated from a Schkuhria species [11]. However, in this case an unseparable mixture of four esters each with a different chain length was present. Acetylation under mild conditions afforded the diacetate 44. The <sup>1</sup>H NMR data (see Experimental) were nearly identical with those of the known ester [11]. The position of the methoxy group was deduced from the observed shift differences of the aromatic protons in the spectra of 43 and 44, while the position of the ester chain methyl group could not be determined. The structure of the diacetate 6 followed from the <sup>1</sup>H NMR data (Table 1), especially when the Eu(fod)<sub>3</sub>induced shifts were taken into consideration. The observed shifts of the C-4 and C-10 methyl signals were in agreement with an acetoxy group at C-3 and not at C-1, while the position of the hydroxyl at C-11 followed from the <sup>1</sup>H NMR data. The configuration at this centre, however, was not determined. All signals could be assigned, after the addition of Eu(fod)<sub>2</sub>, by spin decoupling (Table 1).

The structure of 7 directly followed from the  $^1H$  NMR spectrum (Table 1). The chemical shift of H-14, which could be assigned by spin decoupling, was obviously shifted downfield and was coupled with the methylene doublet at  $\delta$  3.15. Furthermore, the position of the secondary methyl group  $\alpha$  to the keto group followed from the chemical shift of the corresponding methine proton (2.59 tq). Of the remaining two hydroxyls one had to be placed at C-1 and the second at C-19, as is the case with compounds with a hydroxyl at C-20 a pronounced shift of the 2-H signal was observed. The configuration at C-11 was not determined.

The third diterpene, the triol 8 was obviously closely related to 7. To establish the positions of the hydroxyls, 8 was transformed to the dialdehyde 9 by partial oxidation

<sup>\*</sup>Part 319 in the series "Naturally Occurring Terpene Derivatives". For Part 318 see Bohlmann, F., Zdero, C., King, R. M. and Robinson, H. (1981) *Phytochemistry* 20, 1069.

1614

$$RO$$

$$1 \quad R = Ac$$

$$2 \quad R = H$$

R = OAc, R', R'' = O

18 R. R' = O, R" = OAc

19 R = H, R' = OH, R'' = OAc

30 
$$R = O$$
,  $R', R'' = O$ 

$$R = O$$

$$R' = OH, R'' = H$$

32 
$$R = OH, R' = O$$
,  $R'' = H$ 

Me 
$$(CH_2)_n$$
  $CH(Me)$   $(CH_2)_n$   $CH(OH)$   $CH_2$   $CO_2$   $CH_2$   $CH_2$ 

Table 1. <sup>1</sup>H NMR spectral data of compounds 6-9 (270 MHz, CDCl<sub>3</sub>, TMS as internal standard)

	6*	$6(+\mathrm{Eu}(\mathrm{fod})_3)$	7	8	9
H-1		2.59 d (br)	4.09 dq	4.07 d (br)	9.91 d
H-1′		$2.12 \ dd \ (br)$	4.05 ug	4.07 & (01)	J.J1 u
H-2		2.71 ddd	5.48 t (br)	5.47 t (br)	5.97 d (br)
H-2'		$3.36 \ d \ (br)$	3.40 (01)	3.47 1 (01)	3.77 a (61)
H-3	4.52 dd	7.73 (br)			_
H-4	_	- }	22-20 m	2.3-2.1 m	2.79 m
H-5		2.02 dd }	2.2-2.0 //	2.3-2.1 m }	2.19 m
H-6 }	1.94 m	2.51 dd (br)	5.19 t (br)	5.40 t (br)	6.41 t (br)
H-6′∫	1.74 111	$2.38 \ d \ (br)$	3.19 t (Ur)		
H-7	$5.58 \ s \ (br)$	5.96 d (br)			
H-8			$2.2-2.0 \ m$	2.3-2.1 m	2.27 t (br)
H-9	$2.13 \ s \ (br)$	$3.61 \ s \ (br)$	1.4 m	2.3-2.1 m	2.14 dt
H-10			1.6 m	$5.33 \ t \ (br)$	5.35 t (br)
H-11	4.08 dd (br)	6.08 m	2.59 tq	_	
11 12	2.92	∫ 3.76 dd		200 + (1-)	207 + (1-)
H-12	2.82 m	(3.15 d (br))		$3.99 \ t \ (br)$	$3.97 \ t \ (br)$
H-13		_	3.15 d (br)	2.3-2.1 m	2.25 d (br)
H-14	5.48 t (br)	7.38 (br)	5.19 (br)	5.10 m	5.09 t (br)
H-15	4.61 d (br)	∫ 7.18 dd		_	
11 15	4.01 4 (01)	7.11 dd			
H-16	1.76 s (br)	$2.56 \ s \ (br)$	$1.63 \ s \ (br)$	$1.74 \ s \ (br)$	$1.73 \ s \ (br)$
H-17	$1.82 \ s \ (br)$	$2.32 \ s \ (br)$	$1.57 \ s \ (br)$	$1.66 \ s \ (br)$	$1.65 \ s \ (br)$
H-18	$0.92 \ s$	1.83 s	1.09 d	$1.64 \ s \ (br)$	1.61 s (br)
H-19	0.94 s	1.76 s	$4.08 \ s \ (br)$	4.08 s	10.09 s
H-20	0.87 s	1.66 s	$1.76 \ s \ (br)$	$1.74 \ s \ (br)$	2.02 d
OAc	2.05 s	4.34 s			
	2.06 s	4.20 s			

<sup>\* 400</sup> MHz.

J(Hz): 6: 1,1' = 13; 1',2 = 13; 2,3 = 11; 2',3 = 5; 5,6 = 12; 5,6' = 3; 6',7 = 4; 11,12 = 10; 11,12' = 2; 12,12' = 14; 14,15 = 7; 7: 1,2 = 5,6 = 10,11 $= 13.14 \sim 7$ ; 8/9: 1,2 = 5,6 = 8,9 = 9,10 = 13,14 = 7; 2,20 = 1.5.

with manganese dioxide. Careful spin decoupling allowed the assignment of all signals. As the proton at the hydroxyl-bearing carbon in **9** showed an allylic coupling with the olefinic proton, which itself was coupled with one methyl only, the hydroxyl could only be placed at C-8 or C-12. Spin decoupling showed that the aldehyde group had to be placed at C-1 or C-19. This, however, excluded a C-8 position of the hydroxyl. Its configuration was not determined. The stereochemistry of the double bonds followed from the <sup>1</sup>H NMR shifts of the olefinic protons.

The structures of the costic acid derivatives 10 and 11, which could only be separated by HPLC, followed from the  $^1$ H NMR data (Table 2), which were very similar to those of the methyl ester of costic acid. The main differences in the spectra of 10 and 11 were in the signals of H-7, which were only in agreement with an equatorial isopropenyl group in 10 and an axial one in 11. The position of the hydroxyl group followed from spin decoupling and the chemical shifts of H-15, which excluded a 3-position, while from the couplings of H-1 the  $\beta$ -orientation of the hydroxyl could be deduced in both isomers. The CD curves of 10 and 11 were opposite, as expected (10 negative and 11 positive Cotton effect). As costic acid showed a negative Cotton effect, the absolute configurations were also established.

The structure and stereochemistry of the eudesmanolide 12 were assigned by careful spin decoupling in deuteriobenzene (Table 2). The  $1\beta$ -configuration of the hydroxyl group followed from the couplings and by irradiation of the multiplet at  $\delta$  1.40, which caused the

Table 2. <sup>1</sup>H NMR spectral data of compounds 10–12 (170 MHz, CDCl<sub>3</sub>)

	10	11	12	12(C <sub>6</sub> D <sub>6</sub> )
Η-1α	3.45 d (br)	3.40 dd (br)	3.56 dd	2.92 dd
H-2	1.56 m	1.25-1.9 m		1.40 m
H-3α	2.13 ddd (br	) 2.09 ddd (br)		1.60 ddd
$H-3\beta$	2.32 ddd	2.3 ddd	2.37 ddd	1.95 ddd
H-5		)		1.35 m
H-6x }	1.99-1.3 m	1.25–1.9 $m$	1-1	
H-6β)		)	3.99 dd	3.36 dd
H-7	2.53 dddd	$3.07 \ s \ (br)$		1.40 m
H-8α	)	1		
Η-8β (	100 12	→ 1.25–1.9 m	4.01 ddd	3.43 ddd
H-9a (	1.98-1.5 m	7 1.23-1.9 m	1.44 dd	1.02 dd
H-9 $\beta$	J		2.51 dd	2.27 dd
H-11			2.58 dq	2.08 dq
H-13	6.17 d	$6.19 \ s \ (br)$	1.20	1.52.1
H-13'	5.57 dd	$5.64 \ d \ (br)$	1.39 d	1.53 d
H-14	$0.72 \ s$	0.75 s	$0.83 \ s$	0.54 s
H-15	4.76 ddd	$4.77 \ s \ (br)$	$5.08 \ s \ (br)$	4.71 s (br)
H-15'	4.48 ddd	$4.51 \ s \ (br)$	$4.73 \ s \ (br)$	4.39 s (br)
OMe	3.76 s	3.75 s		_ ` ´

 $J(\text{Hz}): \mathbf{10}: 1\alpha, 2\beta = 12; 2\alpha, 3\alpha = 5.5; 2\alpha, 3\beta = 2.5; 2\beta, 3\alpha = 13.5; 2\beta, 3\beta = 5.5; 3\alpha, 3\beta = 14; 3\alpha, 15 = 3\alpha, 15' = 5\alpha, 15 = 5\alpha, 15' = 1.5; 6\alpha, 7\alpha = 7\alpha, 8\alpha = 3.5; 6\beta, 7\alpha = 7\alpha, 8\beta = 12; 7\alpha, 13' = 1; 13, 13' = 1; 15, 15' = 1.5; \mathbf{11}: 1\alpha, 2\alpha = 5; 1\alpha, 2\beta = 12; 2\alpha, 3\alpha = 5.5; 2\alpha, 3\beta = 2.5; 2\beta, 3\alpha = 13.5; 2\beta, 3\beta = 5.5; 3\alpha, 3\beta = 14; 7\beta, 13' = 2; \mathbf{12}: 1\alpha, 2\alpha = 5; 1\alpha, 2\beta = 12; 2\alpha, 3\beta = 5.5; 3\alpha, 3\beta = 13.5; 5\alpha, 6\beta = 6\beta, 7\alpha = 9.5; 7\alpha, 8\beta = 8\beta, 9\alpha = 11.5; 7\alpha, 11\beta = 12; 8\beta, 9\beta = 4; 11\beta, 13 = 7.$ 

expected changes of the H-1 and H-3 signals. Similarly, irradiation of the H-8 and H-13 signals allowed the assignments of the signals of H-6 through H-9 and of H-11. The observed larger couplings of H-8 and H-11 required a  $\beta$ -orientation of these hydrogens. 12, therefore, was 8 $\alpha$ -hydroxy-11 $\beta$ ,13-dihydroreynosin.

The structures of the germacradienolides 15-17 were deduced from the <sup>1</sup>H NMR data (Table 3) of the natural compounds and some of their derivatives. 17 is closely related to similar lactones with an  $8\beta$ -2-methylbutyryloxy residue [12] and the corresponding angelate [13]. Both obviously had the same stereochemistry at all centres. The critical centre was that at C-9, as from models a clear decision was difficult. We, therefore, oxidized 17 to the corresponding ketone 18, which on sodium borohydride reduction afforded the 9-epimer 19 as well as the 11,13dihydro compound 20 with unknown stereochemistry at C-11. Comparison of the <sup>1</sup>H NMR spectra of 17 and 19 showed that in 19 the 9-hydroxy group was  $\alpha$ -orientated as the H-7 signal was shifted downfield. The <sup>1</sup>H NMR data of the ketone 18 showed that a conformation had to be assumed, where the keto group was not in plane with the 1,10-double bond (chemical shift of H-1), and was most probably below the plane. This probably was the reason for the unexpected  $\beta$ -attack of boranate, which normally should be hindered by the  $8\beta$ -ester residue. The configuration of the angelate [13], therefore, must be corrected also to  $9\beta$ -hydroxy. As the spectral data and the couplings in the spectra of 15 and 16 were nearly the same as those of 17, these lactones obviously had the same stereochemistry too.

21–24 were rearranged guaianolides. The <sup>1</sup>H NMR data (Table 4) of 21 showed the typical signals of a tiglate [6.81 qq, 1.81 d (br), 1.82 s (br)]. A singlet at  $\delta$  9.48 indicated an aldehyde group, while typical signals of a trans-fused methylene lactone were present too. Spin decouplings allowed the assignment of the signals for H-1 through H-8. The small value of  $J_{7.8}$ required an α-proton at C-8, while the orientation of the aldehyde group was deduced from the downfield shift of H-8, especially if compared with that in the spectrum of 22. The downfield shift of the H-7 signal in the spectrum of 22 supported the proposed stereochemistry, which was further indicated by the downfield shift of the H-1 $\alpha$  signal in the spectrum of 21. The <sup>13</sup>C NMR of 22 was in agreement with the proposed structure Experimental). The spectrum of 23 clearly showed that the hydroxy group of the ester residue was acetylated. Consequently, the signals of these two lactones were nearly identical. The presence of additional oxygen functions in 24 was deduced from the mass spectrum and from the altered <sup>1</sup>H NMR signals compared with those of 21. The position of the keto group followed from the downfield shift of the H-1 signal, while the  $\alpha$ -orientation of a 3,4-epoxide was deduced from the downfield shift of H-21-24 most probably were formed by a Wagner-Meerwein rearrangement of a guaianolide of type A, present as tiglate 26 in the plant (see below). We have named 21 without an aldehyde function at C-9 (numbering as in a guaianolide) lasiolaenolide. A compound with the same carbon skeleton has been prepared by rearrangement via a guaianolide 9,10epoxide [14].

The guaianolides 25-27 and 29-33 were all derived from the so far unknown  $\Delta$  3,4-isomer of dehydrocostus lactone, for which we propose the name lasiolaenin. The

Table 3. <sup>1</sup>H NMR spectral data of compounds 15-20 (270 MHz, CDCl<sub>3</sub>)

	15	16	17	18	19	20
H-1	4.95 d (br)	5.08 d (br)	5.13 d (br)	5.58 d (br)	5.41 dd (br)	5.38 d (br)
Η-2α	2.59 ddd (br)	(	2.62 ddd (br)	2.78 ddd	2.67 ddd (br)	2.65 m
Η-2β	2.36 ddd -	{1.95-1.6 m	2.41 ddd	2.39 ddd	2.44 ddd	2.3-2.5 m
Η-3α	5.26 dd (br)	(	5.25 dd (br)	5.27 dd	5.30 dd (br)	5.28 dd (br)
$H-3\beta$			_	***	_	_
H-5	$4.95 \ d \ (br)$	4.73 d (br)	$4.87 \ d \ (br)$	$4.92 \ d \ (br)$	$4.93 \ d \ (br)$	4.83 d (br)
Η-6β	5.18 dd	5.14 dd	5.15 dd	5.26 dd	5.13 dd	5.04 dd
Η-7α	2.91 dddd	3.02 dddd	3.00 ddd	3.05 dddd	3.33 ddd	2.3-2.5 m
Η-8α	5.78 ddd (br)	5.97 s (br)	$5.96 \ s \ (br)$	5.57 d	5.70 d (br)	5.39 d
Η-9α	2.32 dd	4.33 dd (br)	$4.31 \ d \ (br)$		_	
Η-9β	2.86 dd (br)	_		_	$4.51 \ dd \ (br)$	$4.47 \ s \ (br)$
H-11					_	2.3-2.5 m
H-13	6.33 d	6.33 d	6.36 d	6.37 d	6.32 d	
H-13'	5.64 d	5.70 d	5.73 d	5.63 d	$5.67 d$ }	1.3 d
H-14	$1.52 \ s \ (br)$	$1.54 \ s \ (br)$	$1.59 \ s \ (br)$	$1.71 \ s \ (br)$	$1.51 \ s \ (br)$	$1.51 \ s \ (br)$
H-15	1.78 d	1.79 d	1.80 d	1.84 d	1.78 d	1.74 d
ОН		2.77 d (br)			1.88 d	1.80 d
OAc	$2.13 \ s$		2.13 s	2.14 s	2.13 s	2.12 s
OTigl	$1.83 \ s \ (br)$	1.84 s (br)	$1.85 \ s \ (br)$	$1.87 \ s \ (br)$	$1.83 \ s \ (br)$	$1.88 \ s \ (br)$
	1.82 dg	1.83 d (br)	$1.84 \ d \ (br)$	1.87 d (br)	$1.82 \ d \ (br)$	1.86 dq
	6.82 qq	6.88 m	6.88 m	6.91 m	6.84 m	6.91 gg

J(Hz): **15**:  $1.2\alpha = 5$ ;  $1.2\beta = 12$ ;  $2\alpha.2\beta = 12$ ;  $2\alpha.3\alpha = 6$ ;  $2\beta.3\alpha = 10.5$ ;  $5.6\beta = 10$ ; 5.15 = 1.5;  $6\beta.7\alpha = 8.5$ ;  $7\alpha.13 = 3.6$ ;  $7\alpha.13' = 3$ ;  $8\alpha.9\alpha = 2$ ;  $8\alpha.9\beta = 5.5$ ;  $9\alpha.9\beta = 12$ ; **16**:  $5.6\beta = 10$ ; 5.15 = 1.5;  $6\beta.7\alpha = 9$ ;  $7\alpha.8\alpha = 0.5$ ;  $7\alpha.13 = 3.6$ ;  $7\alpha.13' = 3$ ;  $8\alpha.9\alpha = 2$ ;  $9\alpha.9\alpha = 2$ ;  $9\alpha.9\alpha = 2$ ;  $9\alpha.9\alpha = 2$ ; 17:  $1.2\alpha = 5$ ;  $1.2\beta = 12$ ;  $2\alpha.2\beta = 12$ ;  $2\alpha.3\alpha = 6$ ;  $2\beta.3\alpha = 10.5$ ;  $5.6\beta = 10$ ; 5.15 = 1.5;  $6\beta.7\alpha = 8.5$ ;  $7\alpha.8\alpha = 0.5$ ;  $7\alpha.13 = 3.5$ ;  $7\alpha.13' = 3$ ;  $8\alpha.9\alpha = 2$ ; **18**:  $1.2\alpha = 5.5$ ;  $1.2\beta = 12$ ;  $2\alpha.2\beta = 12$ ;  $2\alpha.3\alpha = 5.5$ ;  $2\beta.3\alpha = 10.5$ ;  $5.6\beta = 10$ ; 5.15 = 1.5;  $6\beta.7\alpha = 9$ ;  $7\alpha.8\alpha = 1$ ;  $7\alpha.13 = 3.5$ ;  $7\alpha.13' = 3$ ; 13.13' = 0.6; **19**:  $1.2\alpha = 4.5$ ;  $1.2\beta = 12.5$ ;  $2\alpha.2\beta = 12$ ;  $2\alpha.3\alpha = 6$ ;  $2\beta.3 = 10$ ;  $5.6\beta = 10$ ; 5.15 = 1.5;  $6\beta.7\alpha = 8.5$ ;  $7\alpha.13 = 3.5$ ;  $7\alpha.13' = 3$ ;  $9\alpha.9\beta = 5$ ;  $9\beta.OH = 4$ ; **20**:  $2\alpha.3\alpha = 6$ ;  $2\beta.3\alpha = 10.5$ ;  $5.6\beta = 10$ ; 5.15 = 1.5;  $6\beta.7\alpha = 9$ ;  $8\alpha.9\beta = 5.3$ ;  $9\beta.OH = 3.5$ .

hydrogens at C-1 and C-5 were most probably both aorientated in each lactone, as the couplings were the same as in all known guaianolides of this type [15]. As ligustin, where no configuration at C-1 was reported [16], had been correlated with estafiatin, this lactone also was no exception. Most probably, the configuration at C-1 in the lactones from Stevia setifera [17] will have to be corrected. The presence of 6,12-trans-methylene lactones in all compounds followed from the observed couplings  $J_{5,6}$  and  $J_{6,7}$  (Table 5), while the configuration at C-9 caused problems. Inspection of models indicated that the angles between H-8α and both protons at C-9 were nearly identical. Obviously, the configurations at C-8 and C-9 were identical in the lactones 25-27 and 29-32. Biogemetical considerations would support the H-9a configurations, as in the lactones 15–17. The differences in the <sup>1</sup>H NMR spectra of 25 and 26 clearly showed that these lactones differed only in the ester part at C-8, while the data of 27 indicated that the 9-hydroxy group was acetylated. The '14 NWR data of 29 showed by the differences of the chemical shifts of H-8 and H-9 and by spin decoupling that these lactones were isomeric. As the chemical shifts were not very different, we had transformed 27 to the corresponding ketone 28, its 3H NMR data establishing the proposed assignments of the relative positions at C3 and C3 in the lactones 25-27. As expected the 8x-H signal was shifted downfield and was coupled with H-7 only.

The <sup>1</sup>H NMR data of 31 and 32 clearly indicated that we were dealing with the epoxides of 25 and 29 respectively, while those of 30 showed that an epoxide of 28 was present. (Table 5). Though the configuration at C-3 and C-4 could not be determined with certainty, αepoxides seemed to be more likely, as the chemical shift of H-5 should be at higher fields in compounds with a  $\beta$ epoxide group. 36, together with 28, was obtained by oxidation of 27 with pyridine dichromate. This further supported the presence of  $\alpha$ -epoxides, as the  $\beta$ -side of 27 was sterically hindered. 33, was obviously the 2-oxo derivative of 27; the interpretation of the <sup>1</sup>H NMR data, however, caused some difficulties. At room temperature, two conformations in a nearly equal ratio were recognized. 'At 'nigher temperature, 'the ratio of these two was changed in favour of the conformer with the H-13 signals more downfield. Inspection of models explained the observed differences in the chemical shifts and couplings of the two conformers in part only. However, 33 As at deministrator was day is while members transformed to 35 (see below). The structures of the lactones 34-41 were deduced directly from the <sup>1</sup>H NMR data (Table 6) together with spin decoupling in the usual way. In the spectrum of 34 the absence of a function at C-9 was easily. deduced from the corresponding H-9 signals, while the thanged sinuarions of the double bond followed from the new olefinic methyl signal in all lactones. The chemical shift of the latter agreed with a cisoid arrangement to the

Table 4. <sup>1</sup>H NMR spectral data of compounds 21-24 (270 MHz, CDCl<sub>3</sub>)

	21	22	23	24
H-1α	2.90 ddd	2.32 ddd	2.32 ddd	3.42 d
Η-2α	2.28 ddq	2.21 dd (br)	2.20 dd (br)	
Η-2β	2.48 dddq	2.46 dd (br)	2.44 dd (hr)	
H-3	5.55 dq	$5.49 \ s \ (hr)$	5.49 s (br)	3.34 s
Η-5α	2.49 dd	2.62 dd (br)	2.63 dd (br)	3.09 dd
Η-6β	4.11 dd	4.16 dd	4.16 dd	3.96 dd
H-7α	2.74 dddd	3.17 dddd	3.19 dddd	3.0  m
Η-8α	6.06 d (br)	$5.84 \ d \ (br)$	$5.89 \ d \ (br)$	5.98 s (br)
11.0	0.40	$3.69 \ d \ (br)$	$3.71 \ d \ (br)$	9.51 s
H-9	9.48 s	$3.52 \ d \ (br)$	$3.53 \ d \ (br)$	9.31 8
H-13	6.15 d	6.11 d	6.12 d	6.26 d
H-13'	5.57 d	5.52 d	5.53 d	5.85 d
H-14	1.10 s	0.99 s	1.02 s	1.23 s
H-15	1.91 ddd	$1.89 \ s \ (br)$	$1.89 \ s \ (br)$	1.80 s
OR	$1.82 \ s \ (br)$	$1.80 \ s \ (br)$	1.98 s	1.95 s
	$1.81 \ d \ (br)$	1.79 d (br)	1.95 d	1.95 d
	6.81 <i>qq</i>	6.77 qq	4.83 d	4.73 s
			4.77 d	7.09 q
			$7.08 \ q$	

J(Hz): 21:  $1\alpha,2\alpha = 8$ :  $1\alpha,2\beta = 11.5$ ;  $1\alpha,5\alpha = 7.5$ ;  $2\alpha,2\beta = 14.5$ ;  $2\alpha,15 = 1$ ;  $2\beta,3 = 1$ ;  $2\beta,15 = 2$ ; 3,15 = 2;  $5\alpha,6\beta = 10.5$ ;  $6\beta,7\alpha = 11.5$ ;  $7\alpha,8\alpha = 2.5$ ;  $7\alpha,13 = 7\alpha,13' = 3$ ; 22:  $1\alpha,2\alpha = 8$ ;  $1\alpha,2\beta = 11$ ;  $1\alpha,5\alpha = 7.5$ ;  $2\alpha,2\beta = 14$ ;  $5\alpha,6\beta = 11$ ;  $6\beta,7\alpha = 11.5$ ;  $7\alpha,8\alpha = 2$ ;  $7\alpha,13 = 3.3$ ;  $7\alpha,13' = 3$ ; 14,14' = 11; 23:  $1\alpha,2\alpha = 8$ ;  $1\alpha,2\beta = 11$ ;  $1\alpha,5\alpha = 7.5$ ;  $2\alpha,2\beta = 14$ ;  $5\alpha,6\beta = 11$ ;  $6\beta,7\alpha = 11.5$ ;  $7\alpha,8\alpha = 2$ ;  $7\alpha,13 = 3.3$ ;  $7\alpha,13' = 3$ ; 14,14' = 11; 24:  $1\alpha,5\alpha = 7$ ;  $5\alpha,6\beta = 6\beta,7\alpha = 11.5$ ;  $7\alpha,13' = 3$ ; 14,14' = 11; 24:  $1\alpha,5\alpha = 7$ ;  $13\alpha,6\beta = 6\beta,7\alpha = 11.5$ ;  $13\alpha,6\beta = 11.5$ ;

keto group at C-2. Furthermore, this situation was supported by the downfield shift of H-5, now being allylic. The configurations at C-3 through C-9 most probably were the same as in the lactones 30-33.

The mass spectra of 39–41 indicated the presence of chlorine, which had to be at C-3, as the <sup>1</sup>H NMR data of H-5 through H-9 were very similar to those of 34–38. Some of the chemical shifts were altered in the expected way only. The presence of the cross-conjugated ketone caused a considerable downfield shift of H-5, H-14 and H-15 and a slight shift for that of H-9. Oxidation of 41 afforded the diketone 42, the <sup>1</sup>H NMR data of which provided further support for the proposed structure. 34–38 were derived from ludartin and 39–41 from dehydroleucodin [17]. Inspection of the <sup>1</sup>H NMR spectrum of the crude polar fraction showed that at least the major compounds were present. Therefore, 34–41 most probably were not artefacts.

The compounds isolated from this Lasiolaena species indicated a close relationship to Agrianthus by the co-occurrence of highly oxygenated guaianolides. Germacranolides similar to 15-17 were isolated from a Grazielia species [18], which, however, is placed in the Disynaphia group.

## EXPERIMENTAL

Air-dried plant material, collected in north-eastern Brazil (voucher RMK 8138), was extracted with Et<sub>2</sub>O-petrol. The extracts were separated first by column chromatography (Si gel,

act. grade II), then by repeated TLC (Si gel, GF 254) and in part by HPLC (reversed phase, MeOH-H2O, 7:3). Known compounds were identified by comparison (IR and <sup>1</sup>H NMR) with authentic material. The roots (50 g) afforded 1 mg squalene, 2 mg dammadienone. 8 mg dammadienyl acetate and 3 mg dammadienol. The less polar fractions of the aerial parts (150g) gave 20 mg germacrene D. 15 mg squalene, 2 mg caryophyllene epoxide, 3 mg dammadienone, 10 mg dammadienyl acetate, 5 mg dammadienol, 10 mg lupevl acetate, 15 mg lupeol, 10 mg 1, 15 mg 2, 3 mg 3, 4 mg 4, and 10 mg 5. The more polar fractions were combined in three parts of slightly different polarity (column chromatography, Et<sub>2</sub>O and Et<sub>2</sub>O-MeOH, 19:1). Each part was further separated first by TLC using CHCl3-acetone, 10:1 and then Et<sub>2</sub>O-petrol, 4:1 (developed several times). 10, 11, 15, 25 and 37 could be obtained pure by HPLC only. Finally, 20 mg 6, 15 mg 7, 400 mg 8, 3 mg 10, 1 mg 11, 10 mg 12, 5 mg 13, 50 mg 14, 5 mg 15, 5 mg 16, 50 mg 17, 5 mg 21, 12 mg 22, 10 mg 23, 10 mg 24, 3 mg 25, 3 mg 26, 30 mg 27, 20 mg 29, 2 mg 30, 10 mg 31, 5 mg 32, 6 mg 33, 15 mg 34, 2 mg 35, 2 mg 36, 100 mg 37, 30 mg 38, 2 mg 39. 2 mg 40, 30 mg 41, 50 mg 43 and 20 mg 45 were obtained.

 $3\alpha$ , 15-Diacetoxy-11-hydroxy-ent-labda-7,13-diene (6). Colourless gum; IR  $v_{\text{max}}^{\text{CHCI}_{8}}$  cm $^{-1}$ : 3600 (OH), 1725, 1260 (OAc); MS m/z (rel. int.): 406.272 (M $^{+}$ , 1) (C<sub>24</sub>H<sub>38</sub>O<sub>5</sub>). 346 (M $^{-}$ HOAc, 1), 278 (M $^{-}$ Me<sub>2</sub>C=CHCH<sub>2</sub>OAc, 12). 250 (278  $^{-}$ CO, 6), 235 (250  $^{-}$ Me, 3), 190 (250  $^{-}$ HOAc, 79). 175 (190  $^{-}$ Me, 44), 97 (100);

$$[\alpha]_{24}^{\lambda} = \frac{589}{+15} \frac{578}{+16} \frac{546}{+19} \frac{436 \text{ nm}}{+31} (c = 0.3, \text{CHCl}_3).$$

19-Hydroxy-12-oxo-10,11-dihydrogeranylgeraniol (7). Colourless gum; IR  $v_{max}^{\rm CHCI_3}$  cm $^{-1}$ : 3610 (OH), 1720 (C=O); MS m/z (rel. int.): 322.251 (M $^+$ , 0.3) (C $_{20}$ H $_{34}$ O $_{3}$ ), 304 (M $_{-}$ H $_{2}$ O, 1), 235 (304 - Me $_{2}$ C=CHCH $_{2}^+$ , 9), 217 (235 - H $_{2}$ O, 9), 189 (217 - CO. 10), 161 (189 - CO, 15), 69 (Me $_{2}$ C=CHCH $_{2}^+$ , 100). 12,19-Dihydroxygeranylgeraniol (8). Colourless gum, IR  $v_{max}^{\rm CCI}$ 

12,19-Dalyaro's gerand's Gerand's (8). Colombies guin, Re  $V_{\text{max}}$  cm<sup>-1</sup>: 3610, 3370 (OH); MS m/z (rel. int.): 307 (M<sup>+</sup> – Me, 0.3), 293 (M – CHO, 0.3), 279 (307 – CO, 0.3), 69 (Me<sub>2</sub>C=CHCH<sub>2</sub><sup>+</sup>, 100), 20 mg 8 in 2 ml Et<sub>2</sub>O were stirred for 2 hr with 100 mg MnO<sub>2</sub>. TLC (Et<sub>2</sub>O-petrol, 1:1) afforded 10 mg 9 as a colourless gum: 1R  $v_{\text{max}}^{\text{CCL}_3}$  cm<sup>-1</sup>: 3620 (OH), 1750, 1685 (C=CCHO); MS m/z (rel. int.): 318.220 (M<sup>+</sup>, 0.5) (C<sub>20</sub>H<sub>30</sub>O<sub>3</sub>), 249 (M – Me<sub>2</sub>C=CHCH<sub>2</sub>, 44), 231 (249 – H<sub>2</sub>O, 46), 213 (231 – H<sub>2</sub>O, 28), 69 (Me<sub>2</sub>C=CHCH<sub>2</sub><sup>+</sup>, 100);

$$[\alpha]_{24}^{2} = \frac{589}{+8} \frac{578}{+9} \frac{546}{+10} \frac{436 \text{ nm}}{+17} (c = 1.2, \text{CHCl}_3).$$

1β-Hydroxycostic acid methyl ester (10). Colourless oil, IR  $v_{\rm max}^{\rm CC1}$  cm<sup>-1</sup>: 3610 (OH), 1730 (C=CCO<sub>2</sub>R); MS m/z (rel. int.): 264.173 (M<sup>+</sup>, 42) (C<sub>16</sub>H<sub>24</sub>O<sub>3</sub>), 246 (M – H<sub>2</sub>O, 100), 232 (M – MeOH, 21), 220 (M – CH<sub>2</sub>=CHOH, 27), 205 (220 – Me, 31), 187 (205 – H<sub>2</sub>O, 29); CD (MeCN) last reading: 198 nm (negative) (costic acid also).

 $1\beta$ -Hydroxy-7-epi-costic acid methyl ester (11). Colourless oil, IR  $v_{\rm max}^{\rm CCL_1}$  cm  $^{-1}$ : 3610 (OH), 1730 (C=CCO<sub>2</sub>R); MS m/z (rel. int.): 264.173 (M $^+$ , 40) (C $_{16}$ H<sub>24</sub>O<sub>3</sub>), 246 (M $_{-}$ H<sub>2</sub>O, 100), 232 (M $_{-}$ MeOH, 18), 220 (M $_{-}$ H<sub>2</sub>O=CHOH, 77), 215 (220  $_{-}$ Me, 31); CD (MeCN) last reading: 195 nm (positive).

8α-Hydroxy-11 $\beta$ ,13-dihydroreynosin (12). Colourless gum, IR  $v_{\rm max}^{\rm CHG_3}$  cm  $^{-1}$ : 3605 (OH). 1770 ( $\gamma$ -lactone); MS m/z (rel. int.): 266.152 (M  $^+$ , 0.2) ( $C_{15}H_{22}O_4$ );

$$[\alpha]_{24}^{\lambda} = \frac{589}{+20} \quad \frac{578}{+22} \quad \frac{546}{+25} \quad \frac{436 \text{ nm}}{+40} - (c = 0.32, \text{ CHCl}_3).$$

 $3\beta$ -Acetoxy-8 $\beta$ -tiglinoyioxy-costunolide (15). Colourless gum; IR  $v_{max}^{CHC1}$ , cm $^{-1}$ : 1765 ( $\gamma$ -lactone), 1740, 1260 (C=CCO $_2$ R); MS

Table 5. <sup>1</sup>H NMR spectral data of compounds 25-34 (270 MHz, CDCl<sub>3</sub>)

	25	26	27	28	29	30	31	32	33	ਲ
	3.02 dd (br)	3.0 m	3.0 m	3.43 ddd	3.0 m	3.11 ddd	3.00 ddd	3.02 ddd	3.57 d	2.98 d
	2.43 d (br)	2.44 d (br)	2.44 d (br)	(227.10.)	2.44 d (br)	2.13 dd	2.19 dd	2.29 dd (br)	1	ļ
	2.70 m	2.67 d (br)	2.68 d (br)	$\left\{ 2.57 \ a \ (br) \right\}$	2.66 d (br)	1.70 ddd	2.32 ddd	1.98 ddd	1	
	5.63 s (br)	5.63 dd (br)	5.64 s (br)	5.54 m	5.59 m	3.36 s	$3.43 \ s \ (br)$	3.46 s (br)	3.37 s	3.41 s
	3.09 d (br)	3.0 m	3.0 m	2.68 dd (br)	3.08 dd (br)	2.46 dd	2.44 dd	2.42 dd	2.93 dd	2.55 dd
	4.44 dd	4.43 dd	4.43 dd	4.25 dd	4.51 dd	4.23 dd	4.56 dd	4.21 dd	5.09 dd	4.60 dd
	3.14 m	3.10 dddd	3.13 dddd	3.72 dddd	2.96 m	3.78 dddd	3.13 dddd	2.90 dddd	3.33 dddd	2.65 dddd
	5.95 dd	5.87 dd	5.94 dd	6.02 d	4.47 s (br)	6.02 d	5.73 dd	4.45 ddd	5.40 dd	5.45 dd
	5.32 d	4.38 dd (br)	4.39 dd (br)	1	5.28 d (br)		4.56 dd	5.25 d (br)	4.86 dd	5.16 d (br)
	6.30 d	6.30 d	6.32 d	6.37 d	6.39 d	6.39 d	6.28 d	6.40 d	6.30 d	6.25 d
	5.65 d	5.66 d	5.72 d	5.61 d	5.62 d	5.62 d	5.55 d	5.63 d	5.48 d	5.44 d
	5.23 s (br)	5.34 s (br)	5.33 s (br)	5.30 s	£ 15 = (L.)	5.29 s	5.26 s (br)	5.24 s (br)	5.40 s	5.41 s
	5.19 s (br)	5.19 s (br)	5.21 s (br)	5.12 d	( a) s cr.c	5.12 s	5.15 s (br)	5.17 d	5.19 s	5.34 s
	1.86 s (br)	1.86 s (br)	1.86 d (br)	$1.92 \ d \ (br)$	1.84 m	1.64 s	1.63 s	1.64 s	1.75 s	1.66 s
НО		2.32 d	2.47 d		2.70 d	1	2.82 d	2.59 d	2.92 s	1
	2.03 s	J	1	1	1	1		l		
	2.03 s	1.77 s (br)	2.0 s	2.02 s	2.06 s	2.02 s	2.0 s	2.07 s	2.01 s	2.01 s
	1.96 d	1.78 d (br)	1.96 d	1.97 d	2.00 d	1.99 d	1.96 d	2.00 d	1.97 d	1.97 d
	4.83 d	6.80 m	4.83 d	4.89 d	4.98 s	4.91 d	4.82 s (br)	4.99 d	4.88 d	4.88 d
	4.73 d	ļ	4.73 d	4.79 d	7.16 9	4.82 d	6.15 q	4.94 d	4.80 d	4.81 d
	7.10 q	ı	7.11 q	7.16 q	1	7.19 q	1	7.26 q	7.14 q	7.17 q

 $5\alpha_{6}\beta_{8} = 9.5$ ;  $6\beta_{7}\alpha = 9$ ;  $7\alpha_{1}13 = 3.5$ ;  $7\alpha_{1}13' = 3$ ;  $8\alpha_{2}9\alpha = 3$ ;  $9\alpha_{2}OH = 2.5$ ; 30:  $1\alpha_{2}\alpha = 8$ ;  $1\alpha_{2}\beta = 11$ ;  $1\alpha_{2}\delta = 8$ ;  $2\alpha_{2}\beta = 14$ ;  $2\beta_{3}\beta = 12$ ;  $5\alpha_{6}\beta = 11$ ;  $7\alpha_13' = 3$ ;  $8\alpha_9\alpha = 3.5$ ;  $9\alpha_1OH = 2.5$ ; 32:  $1\alpha_2\alpha = 7.5$ ;  $1\alpha_12\beta = 7$ ;  $1\alpha_5\alpha = 3$ ;  $2\alpha_3\beta = 2$ ;  $5\alpha_5\beta = 10.5$ ;  $6\beta_17$ ; = 9;  $7\alpha_5\beta = 2.5$ ;  $7\alpha_13 = 3.2$ ;  $7\alpha_13' = 3$ ;  $J(Hz); \mathbf{25}; 1, 2\beta = 1, 5\alpha = 9; 2\alpha, 2\beta = 16.5; 5\alpha, 6\beta = 6\beta, 7\alpha = 9.5; 7\alpha, 8\alpha = 1.5; 7\alpha, 13 = 3.7; 7\alpha, 13' = 3; 8\alpha, 9\alpha = 3; \mathbf{26}; 2\alpha, 2\beta = 17; 5\alpha, 6\beta = 6\beta, 7\alpha = 9.5; 7\alpha, 13' = 3; 2\alpha, 2\alpha = 3; 2\alpha, 2\beta = 17; 2\alpha, 6\beta = 6\beta, 7\alpha = 9.5; 2\alpha, 2\beta = 17; 2\alpha, 6\beta = 6\beta, 7\alpha = 9.5; 2\alpha, 2\beta = 15; 2\alpha, 2\beta = 17; 2\alpha, 2\beta = 17;$  $7\alpha_{3}\alpha_{3}=2$ ;  $7\alpha_{1}13=3$ ;  $7\alpha_{1}8\alpha=2$ ;  $8\alpha_{2}9\alpha=3.5$ ;  $9\alpha_{1}OH=3$ ; 27:  $2\alpha_{2}\beta=16.5$ ;  $5\alpha_{1}6\beta=10$ ;  $6\beta_{1}7\alpha=9$ ;  $7\alpha_{3}8\alpha=2$ ;  $7\alpha_{1}13=3.5$ ;  $7\alpha_{1}13'=3$ ;  $8\alpha_3\alpha = 9\alpha_3OH = 4$ , 28:  $1\alpha_2\alpha = 1\alpha_32\beta = 1\alpha_35\alpha = 9$ ;  $5\alpha_5\beta = 10.5$ ;  $6\beta_57\alpha = 9$ ;  $7\alpha_58\alpha = 4.2$ ;  $7\alpha_513 = 3.7$ ;  $7\alpha_513' = 3$ ; 29:  $1\alpha_53\alpha C 8.5$ ;  $2\alpha_5\beta = 16.5$ ;  $6\beta_17\alpha = 8.5, 7\alpha_8\alpha = 4.2, 7\alpha_413 = 3.5, 7\alpha_413' = 3; 31: 1\alpha_2\alpha = 8; 1\alpha_2\beta = 1\alpha_5\alpha = 8.5, 2\alpha_2\beta = 14; 5\alpha_46\beta = 10.5; 6\beta_17\alpha = 9; 7\alpha_8\alpha = 7\alpha_413 = 3.5; 7\alpha_8\alpha = 7\alpha_8\alpha$  $8\alpha 9\alpha = 8\alpha$ , OH = 4; 33:  $1\alpha,5\alpha = 5\alpha,6\beta = 6\beta,7\alpha = 9.5$ ;  $7\alpha,9\alpha = 5$ ;  $7\alpha,13 = 3.5$ ;  $7\alpha,13' = 3$ ;  $8\alpha,9\alpha = 5.5$ ;  $9\alpha,$ OH = 3; 34:  $1\alpha,5\alpha = 7$ ;  $5\alpha,6\beta = 7.5$ ;  $5\beta,7\alpha = 10.5$ ;  $7\alpha,8\alpha = 4$ ;  $7\alpha,13 = 3.5$ ;  $7\alpha,13' = 3$ ;  $8\alpha,9\alpha = 5$ .

Table 6. <sup>1</sup>H NMR spectral data of compounds 34-42 (270 MHz, CDCl<sub>3</sub>)

	34	35	36	37	38	39	40	41	42
Η-3β	3.41 s (br)	3.45 s (br)	3.40 s (br)	3.42 s (br)	3.41 s (br)				
Η-5α	3.45 d (br)	$3.49 \ d \ (br)$	$3.43 \ d \ (br)$	$3.44 \ d \ (br)$	3.45 d (br)	3.65 d (br)	$3.29 \ d \ (br)$	$3.56 \ d \ (br)$	3.55 dqq
I-6ß	4.10 dd	4.01 dd	3.99 dd	4.02 dd	4.12 dd	4.01 dd	4.50 dd	4.00 dd	4.41 dd
-1-7α	3.09 dddd	3.33 m	3.27 dádá	3.30 ddd	3.10 dddd	3.40 dddd	3.15 d (br)	3.35 dddd	3.30 dddd
I-8α	5.70 d (br)	$5.62 \ s \ (br)$	5.55 d	5.67 d	$5.70 \ d \ (br)$	$5.75 \ s \ (br)$	5.76 dd (br)	5.69 d (br)	5.40 d
<del>1</del> -9α	2.82 dd	5.94 s (br)	4.61 d	4.65 d (br)	2.86 dd	$6.02 \ s \ (hr)$	2.76 d (br)	4.77 d (br)	
1-9ß	2.63 d (br)		_		$2.66 \ d \ (br)$	_ `	2.95 dd (br)	***	
I-13	6.22 d	6.22 d	6.25 d	6.24 d	6.21 d	6.26 d	6.31 d	6.29 d	6.40 dd
I-13'	5.49 d	5.53 d	5.46 d	5.49 d	5.46 d	5.62 d	5.62 d	5.55 d	5.67 dd
1-14	2.23 d (br)	2,23 s	2.22 s	2.24 s	2.23 s	$2.40 \ s \ (br)$	$2.34 \ s \ (br)$	2.40 s (br)	2.41 d
1-15	1.82 s	1.81 s	1.81 s	1.81 s	1.81 s	$2.40 \ s \ (br)$	$2.34 \ s \ (br)$	2.38 s (br)	2.29 d
OAc		2.16 s		W-1000-0-100		2.20 s		and the contract of	
H	······································	1887-07	4.07 d	4.00 d	1.00			4.12 d (br)	***
	1.76 dg	1.96 d	1.76 da	1.97 d	1.94 d	1.97 s	1.96 d	1.97 d	1.99 d
	6.73 gg	4.75 s	6.79 ag	4.74 s	4.74 s	4.77 s (br)	4.75 s (br)	4.76 d	4.81 d
		7.07 a		7.08 a	7.04 g	7.04 q	7.10 a	4.72 d	4.75 d
	prince .							7.06 a	7.12 g

 $J(\text{Hz}): \ \ 34: \ 5\alpha,6\beta = 6\beta,7\alpha = 10.5; \ 5\alpha,14 = 1; \ 7\alpha,8\alpha = 2; \ 7\alpha,13 = 3; \ 7\alpha,13' = 3; \ 7\alpha,8\alpha = 2.5; \ 8\alpha,9\alpha = 6; \ 9\alpha,9\beta = 15; \ 35: \ 5\alpha,6\beta = 6\beta,7\alpha = 10.5; \ 7\alpha,13 = 3.2; \ 7\alpha,13 = 3.2; \ 7\alpha,13' = 2.8; \ 36: \ 5\alpha,6\beta = 6\beta,7\alpha = 10.5; \ 7\alpha,8\alpha = 2; \ 7\alpha,13 = 3.2; \ 7\alpha,13 = 3; \ 9\alpha,\text{OH} = 8.5; \ 37: \ 5\alpha,6\beta = 6\beta,7\alpha = 10.5; \ 7\alpha,8\alpha = 2; \ 7\alpha,13 = 3.2; \ 7\alpha,13 = 3.2; \ 7\alpha,13 = 3; \ 9\alpha,\text{OH} = 7; \ 38: \ 5\alpha,6\beta = 6\beta,7\alpha = 10.5; \ 7\alpha,8\alpha = 2; \ 7\alpha,13 = 7\alpha,13' = 3; \ 8\alpha,9\alpha = 6; \ 9\alpha,9\beta = 15; \ 5\alpha,14 = 1; \ 39: \ 5\alpha,6\beta = 6\beta,7\alpha = 10.5; \ 7\alpha,8\alpha = 1.5; \ 7\alpha,13 = 2.8; \ 7\alpha,13' = 2.6; \ 40: \ 5\alpha,6\beta = 6\beta,7\alpha = 10.5; \ 7\alpha,8\alpha = 1.5; \ 7\alpha,13 = 3.3; \ 7\alpha,13' = 2.8; \ 8\alpha,9\alpha = 4.5; \ 8\alpha,9\beta = 0.5; \ 9\alpha,9\beta = 17; \ 41: \ 5\alpha,6\beta = 6\beta,7\alpha = 10.5; \ 7\alpha,8\alpha = 1.5; \ 7\alpha,13 = 3.3; \ 7\alpha,13' = 3.0; \ 13',13' = 0.7.$ 

m/z (rel. int.): 388.189 (M<sup>+</sup>, 0.3) (C<sub>22</sub>H<sub>28</sub>O<sub>6</sub>), 346 (M - ketene, 1), 246 (346 - TiglOH, 11), 228 (246 - H<sub>2</sub>O, 9), 83 (C<sub>4</sub>H<sub>7</sub>CO<sup>+</sup>, 100).

9 $\beta$ -Hydroxy-8 $\beta$ -tiglinovloxy-costunolide (16). Colourless gum, IR  $\nu_{\rm max}^{\rm CHCl_3}$ , cm  $^{-1}$ : 3600 (OH), 1770 ( $\gamma$ -lactone), 1720 (C=CCO $_2$ R); MS m/z (rel. int.): 346.178 (M $^+$ , 0.6) (C $_2$ 0 $H_2$ 6O $_5$ ), 328 (M - H $_2$ O, 0.5), 246 (M - TiglOH, 5), 228 (246 - H $_2$ O, 6), 83 (C $_4$ H $_7$ CO $^+$ , 100), 55 (83 - CO, 38).

 $3\beta$ -Acetoxy-9 $\beta$ -hydroxy-8 $\beta$ -tiglinoyloxy-costunolide (17). Colourless crystals, mp 207.5° (Et<sub>2</sub>O); IR  $\nu_{\rm max}^{\rm crg}$  cm<sup>-1</sup>: 3610 (OH), 1770 ( $\gamma$ -lactone), 1730 (OAc, C=CCO<sub>2</sub>R); MS m/z (rel. int.): 404.184 (M<sup>+</sup>, 0.5) (C<sub>22</sub>H<sub>28</sub>O<sub>7</sub>), 362 (M — ketene, 0.8), 345 (M — OAc, 3), 262 (362 — TiglOH, 7), 244 (262 — H<sub>2</sub>O, 12), 83 (C<sub>4</sub>H<sub>7</sub>CO<sup>+</sup>, 100);

$$[\alpha]_{24}^{\lambda} = \frac{589}{+130.4} \frac{578}{+136.8} \frac{546}{+159.1} \frac{436 \text{ nm}}{+304.8} (c = 1.5).$$

10 mg 17 in 0.2 ml CH<sub>2</sub>Cl<sub>2</sub> were stirred for 24 hr with 50 mg pyridine dichromate. TLC afforded 6 mg 18, colourless gum. For <sup>1</sup>H NMR see Table 3. To 6 mg 18 in 1 ml MeOH, 10 mg NaBH<sub>4</sub> and, after a min, dil H<sub>2</sub>SO<sub>4</sub> were added. TLC (Et<sub>2</sub>O-petrol, 4:1) afforded 2 mg 19 and 2 mg 20 (<sup>1</sup>H NMR see Table 3).

Lasiolaenolide-9-al (21). Colourless gum; IR  $v_{\text{max}}^{\text{CHCl}_3}$ , cm<sup>-1</sup>: 2740, 1730 (CHO), 1775 ( $\gamma$ -lactone), 1720, 1660 (C=CCO<sub>2</sub>R); MS m/z (rel. int.): 344.162 (M<sup>+</sup>, 8) (C<sub>20</sub>H<sub>24</sub>O<sub>5</sub>), 244 (M – TiglOH, 10), 216 (244 – CO, 6), 83 (C<sub>4</sub>H<sub>2</sub>CO<sup>+</sup>, 100).

Lasiolaenolide-9-ol (22). Colourless gum; IR  $\nu_{\rm max}^{\rm CHCl_5}$ , cm<sup>-1</sup>: 3620 (OH), 1770 ( $\gamma$ -lactone), 1715, 1650, (C=CCO<sub>2</sub>R); MS m/z (rel. int.): 346.178 (M\*, 6) (C<sub>20</sub>H<sub>20</sub>O<sub>5</sub>), 246 (M — TigIOH, 11), 228 (246 — H<sub>2</sub>O, 2), 215 (246 — CH<sub>2</sub>OH, 8), 83 (C<sub>4</sub>H<sub>7</sub>CO<sup>+</sup>, 100); <sup>13</sup>C NMR (CDCl<sub>3</sub>) (C-1—C-15):  $\delta$  50.1, 34.1, 124.9, 143.9†, 47.7†, 81.3, 47.5†, 71.0, 70.8, 42.5, 144.0†, 167.4, 120.0, 18.8, 16.8 († may be interchangeable):

$$[\alpha]_{24}^{2} = \frac{589}{+18.4} = \frac{578}{+19.6} = \frac{546}{+22.5} = \frac{436 \,\mathrm{nm}}{+40.5} (c = 0.5).$$

5'-Acetoxy-lasiolaenolide-9-al (23). Colourless gum; IR v<sup>CHC1</sup>; cm<sup>-1</sup>; 3610 (OH), 1765 (y-lactone), 1720, 1655 (C=CCO<sub>2</sub>R); MS m/z (rel. int.): 404.184 (M<sup>+</sup>, 8) (C<sub>22</sub>H<sub>28</sub>O<sub>2</sub>), 246 (M - RCO<sub>2</sub>H, 17), 228 (246 - H<sub>2</sub>O, 28), 216 (246 - CH<sub>2</sub>O, 78), 99 (MeCH=C(CH<sub>2</sub>OH)CO<sup>+</sup>, 48), 81 (99 - H<sub>2</sub>O<sub>1</sub> 100).

S'-Acetoxy-3α,4α-epoxy-2-oxo-lasiolaenolide-9-al (24). Colourless gum; IR  $v_{max}^{\rm CIIC1_3}$  cm<sup>-1</sup>: 2740, 1730 (CHO), 1775 (γ-lactone), 1730 (C=CCO<sub>2</sub>R, C=O, OAe); MS m/z (rel. int.): 432.142 (M<sup>+</sup>, 0.4) (C<sub>22</sub>H<sub>24</sub>O<sub>9</sub>), 404 (M - CO, 0.5), 389 (404 - Me, 2), 246 (M - CO, RCO<sub>2</sub>H, 17), 141 (RCO<sup>+</sup>, 100), 99 (141 - ketene, 78), 81 (99 - H<sub>2</sub>O, 82).

9 $\beta$ -Acetoxy-8 $\beta$ -[5-acetoxy-tiglinoyloxy]-lasiolaenin (25). Colourless gum; IR  $\nu_{\text{max}}^{\text{CHCI}_3}$ cm<sup>-1</sup>: 1770 (y-lactone), 1740 (OAc, C=CCO<sub>2</sub>R); MS m/z (ref. int.): 444.178 (M<sup>+</sup>, 3) (C<sub>24</sub>H<sub>28</sub>O<sub>8</sub>), 402 (M - ketene, 0.5), 384 (M - HOAc, 2), 342 (384 - ketene, 1), 244 (402 - RCO<sub>2</sub>H, 5), 226 (244 - H<sub>2</sub>O, 38), 141 (RCO<sup>+</sup>, 100), 99 (141 - ketene, 29), 81 (99 - H<sub>2</sub>O, 95).

9 $\beta$ -Hydroxy-8 $\beta$ -tiglinoyloxy-lasiolaenin (26). Colourless gum; IR  $v_{\text{max}}^{\text{CCL}_4}$  cm  $^{-1}$ : 3600 (OH), 1775 ( $\gamma$ -lactone). 1720 (C=CCO<sub>2</sub>R); MS m/z (rel. int.): 344.162 (M $^+$ , 4) (C<sub>20</sub>H<sub>24</sub>O<sub>5</sub>), 244 (M – TiglOH, 5), 226 (244 – H<sub>2</sub>O, 6), 83 (C<sub>4</sub>H<sub>2</sub>CO $^+$ , 100).

9 $\beta$ -Hydroxy-8 $\beta$ -[5-acetoxy-tiglinoyloxy]-lasiolaenin (27). Colourless gum; IR  $\nu_{max}^{\text{CRU}_3}$  cm $^{-1}$ ; 3600 (OH), 1760 ( $\gamma$ -lactone), 1728 (C=CCO<sub>2</sub>R, OAe); MS m/z (rel. int.); 402.168 (M $^+$ , 4) ( $C_{22}H_{26}O_7$ ), 244 (M-RCO $_2$ H, 20), 226 (244 - H $_2$ O, 10), 141 (RCO $_2$ +, 100), 99 (141 - ketene, 34) 81 (99 - H,O, 59);

$$[\alpha]_{24}^{\lambda} = \frac{589}{+15.7} \frac{578}{+15.7} \frac{546}{+19.1} \frac{436 \text{ nm}}{+32.7} (c = 0.9, \text{CHCl}_3).$$

10 mg 27 in 0.2 ml  $CH_2Cl_2$  were stirred for 72 hr with 50 mg pyridine dichromate. TLC (Et<sub>2</sub>O-petrol, 4:1) afforded 1 mg 30. identical with the natural compound (IR and <sup>1</sup>H NMR) and 4 mg 28, colourless gum; MS m/z (rel. int.): 400.152 (M<sup>+</sup>, 10) (C<sub>22</sub>H<sub>24</sub>O<sub>7</sub>), 340 (M - HOAc, 8) (242 (M - RCO<sub>2</sub>, 11), 141 (RCO<sup>+</sup>, 100), 99 (141 - ketene, 25).

8 $\beta$ -Hydroxy-9 $\beta$ -[5-acetoxy-tiglinoyloxy]-lasiolaenin (29). Colourless gum; IR  $\nu_{max}^{\text{CHCl}_3}$  cm $^{-1}$ : 3520 (OH), 1770 ( $\gamma$ -lactone), 1730 (C=CCO<sub>2</sub>R, OAc); MS m/z (rel. int.): 402.168 (M $^+$ , 0.5) (C<sub>22</sub>H<sub>26</sub>O<sub>7</sub>), 244 (M - RCO<sub>2</sub>H, 7), 226 (244 - H<sub>2</sub>O, 5), 141 (RCO $^+$ , 100), 99 (141 - ketene, 22), 81 (99 - H<sub>2</sub>O, 73).

 $3\alpha,4\alpha-Epoxy-9-oxo-8\beta-[5-acetoxy-tiglinoyloxy]-3,4-dihydro-lasiolaenin (30). Colourless gum; IR <math>v_{max}^{\rm CCla}$  cm $^{-1}$ : 1785 ( $\gamma$ -lactone), 1740 (C=CCO $_2$ R, OAc, C=O); MS m/z (rel. int.): 258.189 (M - RCO $_2$ H, 38) (C $_{15}$ H $_{14}$ O $_4$ ), 243 (258 - Me, 6), 141 (RCO $^+$ , 100), 81 (141 - HOAc, 91);

$$[\alpha]_{24^{\circ}}^{k} = \frac{589}{-20} \frac{578}{-21} \frac{546}{-25} \frac{436 \text{ nm}}{-51} (c = 0.1, \text{ CHCl}_3).$$

 $3\alpha,4\alpha$ -Epoxy-9 $\beta$ -hydroxy-8 $\beta$ -[5-acetoxy-tiglinoyloxy]-3,4-dihydro-lasiolaenin (31). Colourless gum; IR  $\nu_{\text{max}}^{\text{CHCl}_3}$  cm $^{-1}$ : 3610 (OH), 1775 (y-lactone), 1735 (OAc, C=CCO<sub>2</sub>R); MS m/z (rel. int.): 260 (M - RCO<sub>2</sub>H, 3), 245 (260 - Me, 3), 141 (RCO $^+$ , 72), 81 (141 - HOAc, 100).

 $3\alpha,4\alpha-Epoxy-8\beta-hydroxy-9\beta-[5-acetoxy-tiglinoyloxy]-3,4-dihydro-lasiolaenin (32). Colourless gum; IR <math>v_{max}^{CHCl_3}$  cm $^{-1}$ : 3610 (OH), 1775 (y-lactone), 1730 (OAc, C=CCO<sub>2</sub>R); MS m/z (rel. int.): 418.163 (M $^+$ , 0.1) (C<sub>22</sub>H<sub>26</sub>O<sub>8</sub>), 260 (M $^-$ RCO<sub>2</sub>H, 6), 141 (RCO $^+$ , 100), 99 (141 $^-$ ketene, 38), 81 (141 $^-$ HOAc, 98).

3α,4α-Epoxy-9β-hydroxy-8β-[5-acetoxy-tiglinoyloxy]-2-oxo-3,4-dihydro-lasiolaenin (33). Colourless gum; IR  $v_{\rm max}^{\rm CHCl_3}$  cm<sup>-1</sup>: 3600 (OH), 1765 (γ-lactone), 1725 (OAc, C=CCO<sub>2</sub>R, C=O); MS m/z (rel. int.): 432.142 (M  $^{+}$ , 3) (C<sub>22</sub>H<sub>24</sub>O<sub>9</sub>), 372 (M – HOAc, 0.5), 274 (M – RCO<sub>2</sub>H, 12), 256 (274 – H<sub>2</sub>O, 12), 141 (RCO $^{+}$ , 100), 99 (141 – ketene, 27), 81 (141 – HOAc, 75). [α]<sub>D</sub> 0 (c=0.5, CHCl<sub>3</sub>). 10 mg 33 in 2 ml CHCl<sub>3</sub> were stirred with 0.5 g SiO<sub>2</sub> for 24 hr. TLC (Et<sub>2</sub>O-petrol, 4:1) afforded 6 mg 37, identical with the natural compound (<sup>1</sup>H NMR).

 $8\beta$ -Tiglinoyloxy-2-oxo-ludartin (34). Colourless gum; IR  $\nu_{\rm max}^{\rm CHCl}$  cm $^{-1}$ : 1770 ( $\gamma$ -lactone), 1715, 1650 (C=O, C=CCO $_2$ R); MS m/z (rel. int.): 358.142 (M $^+$ , 13) (C $_{20}$ H $_{22}$ O $_6$ ), 258 (M $^-$  TiglOH, 53), 83 (C $_4$ H $_7$ CO $^+$ , 100), 55 (83 $^-$  CO, 88);

$$[\alpha]_{24^{\circ}}^{2} = \frac{589}{-83.3} \quad \frac{578}{-88.7} \quad \frac{546}{-107.6} \quad \frac{436 \text{ nm}}{-297}$$

$$(c = 0.55, \text{ CHCl}_{3}).$$

9 $\beta$ -Acetoxy-9 $\beta$ -[5-acetoxy-tiglinoyloxy]-2-oxo-ludartin (35). Colourless gum; IR  $\nu_{\text{max}}^{\text{CHcl}_3}$  cm $^{-1}$ : 1775 ( $\gamma$ -lactone), 1725 (C=CCO $_2$ R, OAc, C=O); MS m/z (rel. int.): 474.153 (M $^+$ , 1) (C $_2$ <sub>4</sub>H $_2$ <sub>6</sub>O $_1$ <sub>0</sub>), 316 (M $^-$  RCO $_2$ H, 3), 256 (316 $^-$  HOAc, 11), 141 (RCO $^+$ , 100), 99 (141 $^-$  ketene, 27), 81 (141 $^-$  HOAc, 76).

8β-[5-Acetoxy-tiglinoyloxy]-2-oxo-ludartin (36). Colourless gum; IR  $v_{max}^{CHCl_3}$  cm<sup>-1</sup>: 1770 (γ-lactone), 1725 (C=CCO<sub>2</sub>R, OAc, C=O); MS m/z (rel. int.): 416.147 (M<sup>+</sup>, 8) (C<sub>22</sub>H<sub>24</sub>O<sub>8</sub>), 258 (M – RCO<sub>2</sub>H, 96), 141 (RCO<sup>+</sup>, 66), 99 (141 – ketene, 51), 81 (141 – HOAc, 100).

9 $\beta$ -Hydroxy-8 $\beta$ -[5-acetoxy-tiglinoyloxy]-2-oxo-ludartin (37). Colourless gum; IR  $\nu_{max}^{\text{CHCl}_3}$  cm $^{-1}$ : 3600 (OH), 1775 ( $\gamma$ -lactone), 1735 (OAc, C=CCO<sub>2</sub>R, C=O); MS m/z (rel. int.): 432.142 (M $^+$ , 1) (C<sub>22</sub>H<sub>24</sub>O<sub>9</sub>), 414 (M - H<sub>2</sub>O, 2), 256 (414 - RCO<sub>2</sub>H, 12), 141 (RCO $^+$ , 100), 99 (141 - ketene, 25), 81 (141 - HOAc, 71).

9 $\beta$ -Hydroxy-8 $\beta$ -tiglinoyloxy-2-oxo-ludartin (38). Colourless gum; IR  $\nu_{\text{max}}^{\text{CHC}_3}$  cm<sup>-1</sup>: 3600 (OH), 1775 (y-lactone), 1720 (C=CCO<sub>2</sub>R, C=O); MS m/z (rel. int.): 374.137 (M<sup>+</sup>, 1)

 $(C_{20}H_{22}O_7)$ , 275 (M – OTigl, 12), 83  $(C_4H_7CO^+, 100)$ , 55 (83 – CO, 82).

 $9\beta$ -Acetoxy-3-chloro- $8\beta$ -[5-acetoxy-tiglinoyloxy]-dehydroleucodin (39). Colourless gum;  $1R \text{ v}_{13}^{\text{CHCI}_3} \text{ cm}^{-1}$ : 1780 (7-lactone), 1735 (OAc, C=CCO<sub>2</sub>R), 1695 (C=CC=O); MS m/z (rel. int.): 494 (0.3) and 492.118 (M<sup>+</sup>, 1) (C<sub>24</sub>H<sub>25</sub>ClO<sub>9</sub>), 336 (7) and 334 (M - RCO<sub>2</sub>H, 23), 294 (21) and 292 (334 - ketene, 75), 141 (RCO<sup>+</sup>, 90), 99 (141 - ketene, 38), 81 (141 - HOAc, 100).

3-Chloro-8β-[5-acetoxy-tiglinoyloxy]-dehydroleucodin (40). Colourless gum; IR  $\nu_{\rm max}^{\rm CHC1}$  cm<sup>-1</sup>: 1790 (γ-lactone), 1730 (α-OAc, C=CCO<sub>2</sub>R); MS m/z (rel. int.): 436 (6) and 434.113 (M<sup>+</sup>, 18) (C<sub>22</sub>H<sub>23</sub>ClO<sub>7</sub>), 278 (8) and 276 (M - RCO<sub>2</sub>H, 20), 141 (RCO<sup>+</sup>, 75), 59 (141 - ketene, 46), 81 (141 - HOAc, 100).

3-Chloro-9β-hydroxy-8β-[5-acetoxy-tiglinoyloxy]-dehydroleucodin (41). Colourless gum; IR  $\nu_{\rm mx}^{\rm CHCl}$ , cm<sup>-1</sup>: 3600 (OH), 1780 (γ-lactone), 1735 (OAc, C=CCO<sub>2</sub>R); 1700 (C=CCO); MS m/z (rel. int.): 452 (3) and 450.108 (M<sup>+</sup>, 10) (C<sub>22</sub>H<sub>23</sub>ClO<sub>8</sub>), 414 (M – HCl, 3), 392 (1) and 390 (M – HOAc, 3), 294 (2) and 292 (M – RCO<sub>2</sub>H, 5), 141 (RCO<sup>+</sup>, 86), 99 (141 – ketene, 36), 81 (141 – HOAc, 100);

$$[\alpha]_{24}^{\lambda} = \frac{589}{-60} \frac{578}{-62.5} \frac{546}{-73} \frac{436 \text{ nm}}{-152.5} \{c = 0.4, \text{ CHCl}_3\}.$$

10 mg 41 on oxidation with pyridine dichromate afforded after TLC (Et<sub>2</sub>O-petrol, 4:1) 5 mg 42, colourless gum. For <sup>1</sup>H NMR see Table 6.

3-Methoxy-4,5-dihydroxy-phenyl propanol ester (43). Colourless gum; IR  $v_{max}^{CCl_4}$  cm<sup>-1</sup>: 3560 (OH), 1730 (CO<sub>2</sub>R); <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  6.43 (d, 2-H, J = 2 Hz), 6.29 (d, 6-H), 2.58 (t, 7-H, J = 7 Hz), 1.93 (tt, 8-H, J = 7, 7 Hz), 4.13 (t, 9-H, J = 7 Hz), 3.87 (s, OMe), 5.27 s (br), 5.24 s (br) and 2.85 (s (br) OH), 2.51 (dd, 3.5 Hz), 2.40 (dd, J = 15, 9 Hz), 4.00 [m,  $COCH_2CH(OH)$ ], 1.27 [s (br),  $-(CH_2)_n$ ], 0.89 (t (br), Me), 0.87 (d, CHMe); MS m/z (rel. int.): 536.408 (M<sup>+</sup>, 10) (C<sub>32</sub>H<sub>56</sub>O<sub>6</sub>), 522  $(C_{31}H_{52}O_6, 7)$ , 508  $(C_{30}H_{52}O_6, 22)$ , 494.361  $(C_{29}H_{50}O_6, 5)$ , 198  $(M - R-CH=C=O, 33), 180 (H_2C=CHCH_2C_6H_3(OH)_2OMe,$ 100), 153 (180 - CH=CH<sub>2</sub>, tropylium cation, 30). 5 mg 43 was stirred overnight with Ac<sub>2</sub>O. After evapn 3 mg 44 were obtained; <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  6.67 (d, 2-H, J = J = 2 Hz), 6.61 (d, 6-H), 2.67 (t, 7-H, J = 7 Hz), 1.98 (tt, 8-H, J = 7, 7 Hz), 4.16 (t, 9-H, T)J = 7 Hz), 3.83 (s, OMe), 2.29 s and 2.27 s (OAc), 2.51 (dd, J = 15, 3.5 Hz), 2.39 (dd, J = 15, 9 Hz),  $3.99 [m, COCH_2CH(OH)]$ , 1.27 $(s (br), CH_2)_n$ , 0.88 (t, (br), Me), 0.86 (d, CHMe).

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